

Product Information

Phosphatase, Alkaline from bovine intestinal mucosa

Ammonium sulfate suspension, $\geq 2,000$ DEA units/mg protein

P5521

Product Description

CAS Registry Number: 9001-78-9

Enzyme Commission (EC) Number: 3.1.3.1

Synonym: alkaline phosphomonoesterase, phosphomonoesterase, glycerophosphatase, alkaline phosphohydrolase, alkaline phenyl phosphatase, orthophosphoric-monoester phosphohydrolase (alkaline optimum)

K_M :

- 1.5×10^{-3} M (*p*-nitrophenyl phosphate)
- 19×10^{-3} M (phosphoenolpyruvate)

Molecular mass:^{1,2} 140–160 kDa

$E_{278}^{1\%} = 7.6$ –10.5

Isoelectric point:³⁻⁵ several isozymes with a pI range of 4.4–5.8

Bovine intestinal alkaline phosphatase is a dimeric, membrane-derived glycoprotein.^{1,2,6} At least three isoforms exist, which typically possess two N-linked and one or more O-linked glycans per monomer.¹ The enzyme requires zinc, and magnesium or calcium divalent ions for activity.³ The enzyme contains approximately 12% carbohydrate (6% hexoses and 6% other neutral sugars).² Each molecule of alkaline phosphatase contains four zinc atoms and four disulfide bridges.² Maximal activity with alkaline phosphatase is achieved in the presence of magnesium.⁷

Alkaline phosphatase has a broad specificity for phosphate esters of alcohols, amines, pyrophosphate, and phenols. It is routinely used to dephosphorylate proteins and nucleic acids.⁸⁻¹⁰ Other applications of alkaline phosphatase include conjugation to antibodies and other proteins for ELISA, Western blotting, and histochemical detection.^{11,12}

Alkaline phosphatase may be used to dephosphorylate the 5'-termini of DNA or RNA to prevent self-ligation. DNA or RNA can also be tagged with radiolabeled phosphate (via T4 polynucleotide kinase) after dephosphorylation with alkaline phosphatase.¹³ Alkaline phosphatase has also been used to dephosphorylate casein and other proteins.^{14,15}

Several publications,^{16,17} theses,^{18,19} and dissertations²⁰⁻²⁴ have cited use of product P5521 in their research protocols.

pH optimum

- The enzyme is most stable in the pH range 7.5–9.5.²
- The pH optimum for enzymatic activity is pH 8–10.
- The pH optimum will change depending upon substrate, substrate concentration, and ionic concentration.⁴
- The enzyme activity for this product is determined at pH 9.8, using a diethanolamine (DEA) buffer enzyme assay.

Substrates^{9,10,25}

Alkaline phosphatase catalyzes the hydrolysis of phosphate monoesters. Substrates that can be hydrolyzed by alkaline phosphatase include:

- *p*-nitrophenyl phosphate
- Phenyl phosphate
- Phenolphthalein phosphate
- α -glycerol phosphate
- β -glycerol phosphate
- 2-phosphorylglycerate, triosephosphate
- Glucose 6-phosphate

- Glucose 1-phosphate
- Fructose 1-phosphate
- Fructose 6-phosphate
- Adenosine 5-phosphate
- Adenosine 3-phosphate
- Phosphoenolpyruvate
- β -nicotinamide adenine dinucleotide phosphate

Inhibitors^{5,10}

- Chelating agents
- Arsenate
- Cysteine
- Iodine
- Inorganic phosphate
- Pyrophosphate
- Diisopropyl phosphate
- Triphenylphosphate
- Diisopropyl fluorophosphate
- L-phenylalanine

Levamisole (such as Cat. No. L9756) is typically used to inhibit endogenous alkaline phosphatase activity, while only slightly inhibiting the intestinal enzyme.^{26,27}

Precautions and Disclaimer

For R&D use only. Not for drug, household, or other uses. Please consult the Safety Data Sheet for information regarding hazards and safe handling practices.

Product

This product is supplied as an ammonium sulfate suspension, in 3.2 M ammonium sulfate containing 1 mM magnesium chloride (MgCl_2) and 0.1 mM zinc chloride (ZnCl_2), pH 7.0.

Specific Activity: $\geq 2,000$ DEA units/mg protein

Unit Definition: One DEA unit will hydrolyze 1 μmole of 4-nitrophenyl phosphate per minute at pH 9.8 at 37 °C. Diethanolamine (DEA) units are measured in a 1.0 M DEA buffer (pH 9.8) containing 0.5 mM MgCl_2 , with a substrate concentration of 15 mM.

Storage/Stability

Store the product, as supplied, at 2-8 °C. **Do not freeze ammonium sulfate suspensions.**

Preparation Instructions

Dilute solutions of alkaline phosphatase should be prepared in 10 mM Tris HCl (pH 8.0), 1-5 mM MgCl_2 , and 0.1-0.2 mM ZnCl_2 . 50% glycerol can be included for long term storage at 2-8 °C.¹²

References

1. Neumann, H., and Lustig, A., *Eur. J. Biochem.*, **109(2)**, 475-480 (1980).
2. Fosset, M. *et al.*, *Biochemistry*, **13(9)**, 1783-1788 (1974).
3. Besman, M., and Coleman, J.E., *J. Biol. Chem.*, **260(20)**, 11190-11193 (1985).
4. Latner, A.L. *et al.*, *Enzymologia*, **40(1)**, 1-7 (1971).
5. Lazdunski, M. *et al.*, *Can. J. Chem.*, **43(8)**, 2222-2235 (1965).
6. Hsu, H.H. *et al.*, *J. Biol. Chem.*, **260(3)**, 1826-1831 (1985).
7. Morton, R.K., *Biochem. J.*, **60(3)**, 573-582 (1955).
8. Fernley, H.N., "Mammalian Alkaline Phosphatases", in *The Enzymes* (Boyer, P.D., ed.), Vol. IV ("Hydrolysis"), pp. 417-447 (1971).
9. Morton, R.K., *Biochem. J.*, **61(2)**, 232-240 (1955).
10. Morton, R.K., *Biochem. J.*, **61(2)**, 240-244 (1955).
11. Harlow, E., and Lane, D., *Antibodies: A Laboratory Manual*. Cold Spring Harbor Laboratory Press (Cold Spring Harbor, NY), Chapter 9, p. 349 (1988).
12. O'Sullivan, M.J., and Marks, V., *Meth. Enzymol.*, **73(Pt B)**, 147-166 (1981).
13. Maunders, M.J., *Methods Mol. Biol.*, **16**, 331-341 (1993).
14. Kalan, E.B., and Telka, M., *Arch. Biochem. Biophys.*, **85(1)**, 273-275 (1959).
15. Green, M.R. *et al.*, *Anal. Biochem.*, **56(1)**, 43-51 (1973).

16. Webby, G.N., and Close, R.C., *New Zealand Journal of Crop and Horticultural Science*, **19(2)**, 167-175 (1991).
17. Hüttmann, G. *et al.*, *Med. Laser Appl.*, **17**, 9-14 (2002).
18. Greeley, Mary Lyn, "Effects of Diet and Coprophagy on the Body Pool Size and Balance of Pantothenic Acid in Rats". Michigan State University, M.S. thesis, p. 32 (1990).
19. House, Megan A., "Molecular Studies of 5-Azacytidine-Induced Early-Flowering Lines of Flax". Wilfred Laurier University, M.Sc. thesis, p. 122 (2010).
20. Dickinson, George David, "Ligand-Gated Calcium Mobilisation in Higher Plants". University of York, Ph.D. dissertation, p. 98 (2002).
21. Ponnada, Srikanth, "Glycine nitrosation and signature mutations in the p53 tumour suppressor gene: a molecular link between diet and cancers of the gastro-intestinal tract". The Open University, Ph.D. dissertation, p. 60 (2007).
22. Glaros, Elias Nicholas, "The potential of sphingolipid depletion for the treatment of atherosclerosis". University of New South Wales, Ph.D. dissertation, p. 81 (2010).
23. Arias, Sandra Carolina Londoño, "Ethnogeology at the Core of Basic and Applied Research: Surface Water Systems and Mode of Action of a Natural Antibacterial Clay of the Colombian Amazon". Arizona State University, Ph.D. dissertation, p. 232 (2016).
24. Haedicke, Inga Elisabeth, "The Development of Mn^{III}-Porphyrins as T₁ Contrast Agents for Biomedical Applications". University of Toronto, Ph.D. dissertation, p. 130 (2016).
25. Stagni, N. *et al.*, *Biochim. Biophys. Acta*, **761(3)**, 246-251 (1983).
26. van Belle, H., *Biochim. Biophys. Acta*, **289(1)**, 158-168 (1972).
27. Chappellet-Tordo, D. *et al.*, *Biochemistry*, **13(9)**, 1788-1795 (1974).

Notice

We provide information and advice to our customers on application technologies and regulatory matters to the best of our knowledge and ability, but without obligation or liability. Existing laws and regulations are to be observed in all cases by our customers. This also applies in respect to any rights of third parties. Our information and advice do not relieve our customers of their own responsibility for checking the suitability of our products for the envisaged purpose.

The information in this document is subject to change without notice and should not be construed as a commitment by the manufacturing or selling entity, or an affiliate. We assume no responsibility for any errors that may appear in this document.

Technical Assistance

Visit the tech service page at [SigmaAldrich.com/techservice](https://www.sigmaaldrich.com/techservice).

Standard Warranty

The applicable warranty for the products listed in this publication may be found at [SigmaAldrich.com/terms](https://www.sigmaaldrich.com/terms).

Contact Information

For the location of the office nearest you, go to [SigmaAldrich.com/offices](https://www.sigmaaldrich.com/offices).

The life science business of Merck operates as MilliporeSigma in the U.S. and Canada.

Merck and Sigma-Aldrich are trademarks of Merck KGaA, Darmstadt, Germany or its affiliates. All other trademarks are the property of their respective owners. Detailed information on trademarks is available via publicly accessible resources.

© 2022 Merck KGaA, Darmstadt, Germany and/or its affiliates. All Rights Reserved.
P5521pis Rev 04/22 TMG,GCY

