

SIGMA QUALITY CONTROL TEST PROCEDURE

ProductInformation

Enzymatic Assay of β-N-ACETYLGLUCOSAMINIDASE (EC 3.2.1.30) from Jack Beans Sigma Prod. No. A-2264

PRINCIPLE:

PNP-NAG + H_2O $\frac{\beta-N-Acetylglucosaminidase}{}$ p-Nitrophenol + NAG

Abbreviations:

PNP-NAG = p-Nitrophenyl N-Acetyl- β -D-Glucosaminide NAG = N-Acetyl- β -D-Glucosamine

CONDITIONS: $T = 25^{\circ}C$, pH = 5.0, A_{400nm} , Light path = 1 cm

METHOD: Spectrophotometric Stop Rate Determination

REAGENTS:

A. 100 mM Citrate Buffer with 200 mM Sodium Chloride and 0.02% (w/v) Bovine Serum Albumin, pH 5.0 at 25°C
 (Prepare 100 ml in deionized water using Citric Acid, Free Acid, Monohydrate, Sigma Prod. No. C-7129, Sodium Chloride, Sigma Prod. No. S-9625, and Albumin, Bovine, Sigma Prod. No. A-4503. Adjust to pH 5.0 at 25°C with 1 M NaOH.)

- B. 10 mM p-Nitrophenyl N-Acetyl-β-D-Glucosaminide Solution (PNP-NAG) (Prepare 5 ml in deionized water using p-Nitrophenyl N-Acetyl-β-D-Glucosaminide, Sigma Prod. No. N-9376.)
- C. 200 mM Borate Buffer, pH 9.8 at 25°C (Prepare 100 ml in deionized water using Boric Acid, Sigma Prod. No. B-0252. Adjust to pH 9.8 at 25°C with 1 M NaOH.)
- D. β -N-Acetylglucosaminidase Enzyme Solution (Immediately before use, prepare a solution containing 0.05 0.1 unit/ml of β -N-Acetylglucosaminidase in cold Reagent A.)

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PROCEDURE:

Pipette (in milliliters) the following reagents into suitable containers:

	<u>Test</u>	<u>Blank</u>
Reagent A (Buffer) Reagent B (PNP-NAG)	0.40 0.50	0.40 0.50
Mix by inversion and equilibrate to 25°C. Then add:		
Reagent D (Enzyme Solution)	0.10	
Mix by inversion and incubate for exactly 10 minutes at 2	5°C. Then add:	
Reagent C (Borate Buffer) Reagent D (Enzyme Solution)	3.00	3.00 0.10

Mix by inversion and transfer to suitable cuvettes. Record the A_{400nm} for both the Test and Blank using a suitable spectrophotometer.

CALCULATIONS:

Units/ml enzyme =
$$\frac{(\Delta A_{400nm} \text{ Test - } \Delta A_{400nm} \text{ Blank})(4)(df)}{(10)(18)(0.1)}$$

4 = Total volume (in milliliters) of Assay

10 = Time of assay (in minutes) as per the Unit Definition

18 = Millimolar extinction coefficient of p-Nitrophenol at 400 nm¹

0.1 = Volume (in milliliter) of enzyme used

Units/mg solid =
$$\frac{\text{units/ml enzyme}}{\text{mg solid/ml enzyme}}$$

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UNIT DEFINITION:

One unit will hydrolyze 1.0 μ mole of p-nitrophenyl N-acetyl- β -D-glucosaminide to p-nitrophenol and N-acetyl- β -D-glucosamine per minute at pH 5.0 at 25°C.

FINAL ASSAY CONCENTRATION:

In a 1.00 ml reaction mix, the final concentrations are 50 mM citric acid, 100 mM sodium chloride, 0.01% (w/v) bovine serum albumin, 5 mM p-nitrophenyl N-acetyl- β -D-glucosaminide and 0.005 - 0.01 unit β -N-acetylglucosaminidase.

REFERENCES:

Bessey, O.A., Lowry, O.H., and Brock, M.J., (1946) Journal of Biological Chemistry 164, 321-329

Li, S.-C. and Li, Y.-T. (1970) Journal of Biological Chemistry 245, 5153-5160

Borooah, J., Leaback, D.H. and Walker, P.G. (1961) Biochemical Journal 78, 106-110

NOTES:

- 1. The millimolar extinction coefficient is described in Bessey, O.A. et al. (1946).
- 2. This assay is based on the cited references.

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